Hemoglobin

Hemoglobin is a metalloprotein in blood that carries oxygen from the alveolus to all tissues in the body. It is composed of four iron heme subunits that can each individually bind oxygen. These four hemes participate in cooperative binding, which means that when one heme subunit binds oxygen, the others are more likely to bind more oxygen. This gives the oxygen saturation curve its characteristic shape. There are many hemoglobin molecules in red blood cells because the body has high oxygen demands. In areas of high oxygen concentration, the propensity to bind oxygen increases. Such areas include the alveolus, where hemoglobin needs to bind oxygen in order to deliver it to cells. In areas with high carbon dioxide levels, the propensity to bind oxygen decreases, allowing hemoglobin to drop oxygen off in high metabolic areas (because CO2 is a byproduct of metabolism). Additionally, high temperatures and low pH levels reduce oxygen binding. There is a derivative of hemoglobin, myoglobin, which is only found in muscle tissue. It only has one subunit rather than four.



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Hemoglobin Has Four Iron Hemes

He-man-globe with (4) Fork Iron

Hemoglobin is composed of four iron heme subunits that can each individually bind oxygen. These four hemes participate in cooperative binding, which means that when one heme subunit binds oxygen, the others are more likely to bind.

Lots Of Hemoglobin In Red Blood Cells

Lots of He-man-globes in RBC truck

Lots of hemoglobin is needed in red blood cells because of the high oxygen demands of the body. Several factors influence hemoglobin's affinity for oxygen, including pH, carbon dioxide levels, temperature, and 2,3-bisphosphoglycerate (2,3-BPG) concentrations. In summary, the high concentration of hemoglobin in red blood cells is vital for meeting the body's oxygen requirements, especially during increased physical activity or metabolic stress.

High O2 Levels Increase O2 Binding

Lots of O2-tanks cause up-arrow He-man-globe binding

High O2 levels increases hemoglobin's binding affinity for oxygen, allowing it to pick up oxygen in areas like the alveolus. The heightened affinity facilitates the efficient loading of oxygen onto hemoglobin, ensuring that oxygen can be transported effectively to peripheral tissues. This mechanism is crucial for maintaining adequate oxygen delivery throughout the body.

High Temperature Reduces O2 Binding

Hot-flame makes He-man-globe to drop Down-arrow O2-tank

High temperature also reduces oxygen-binding affinity in hemoglobin as part of the physiological response to hyperthermia. This phenomenon is particularly evident during exercise, where muscle activity generates heat, leading to a rightward shift in the oxygen-hemoglobin dissociation curve. This shift enhances oxygen unloading in metabolically active tissues. Conversely, lower temperatures increase hemoglobin's oxygen affinity, promoting oxygen loading in the lungs. These temperature-induced changes are part of the body's adaptive mechanisms to meet varying oxygen demands.

High CO2 Levels Reduce O2 Binding

Many CO2-molecules pushing Down-arrow O2-tanks Off

High carbon dioxide levels reduce hemoglobin's binding affinity for oxygen, allowing it to drop oxygen off at tissues. Because tissues are undergoing cellular metabolism, they produce CO2 as a byproduct and use O2 as an electron acceptor.

Low pH Reduces O2 Binding

Acidic-lemon with Down-arrow dropping O2 tanks

Low pH reduces O2 binding in a condition called acidosis. If respiration levels are low and CO2 is not being exhaled fast enough, blood becomes acidic and O2 binding in hemoglobin is reduced.

Myoglobin Binds Oxygen in Muscle

Mayo-globe held by Muscle-man

Myoglobin is an oxygen-binding protein found exclusively in muscle cells, including skeletal and cardiac muscles. Unlike hemoglobin, which is composed of four subunits, myoglobin consists of a single polypeptide chain. This structure enables myoglobin to bind oxygen molecules with high affinity, facilitating efficient oxygen storage and transport within muscle tissues.

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