

Hemoglobin

Hemoglobin is a metalloprotein in blood that carries oxygen from the alveolus to all tissues in the body. It is composed of four iron heme subunits that can each individually bind oxygen. These four hemes participate in cooperative binding, which means that when one heme subunit binds oxygen, the others are more likely to bind more oxygen. This gives the oxygen saturation curve its characteristic shape. There are many hemoglobin molecules in red blood cells because the body has high oxygen demands. In areas of high oxygen concentration, the propensity to bind oxygen increases. Such areas include the alveolus, where hemoglobin needs to bind oxygen in order to deliver it to cells. In areas with high carbon dioxide levels, the propensity to bind oxygen decreases, allowing hemoglobin to drop oxygen off in high metabolic areas (because CO₂ is a byproduct of metabolism). Additionally, high temperatures and low pH levels reduce oxygen binding. There is a derivative of hemoglobin, myoglobin, which is only found in muscle tissue. It only has one subunit rather than four.



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Hemoglobin Has Four Iron Hemes

[He-man-globe with \(4\) Fork Iron](#)

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Lots Of Hemoglobin In Red Blood Cells

[Lots of He-man-globes in RBC truck](#)

Lots of hemoglobin is needed in red blood cells because of the high oxygen demands of the body. Several factors influence hemoglobin's affinity for oxygen, including pH, carbon dioxide levels, temperature, and 2,3-bisphosphoglycerate (2,3-BPG) concentrations. In summary, the high concentration of hemoglobin in red blood cells is vital for meeting the body's oxygen requirements, especially during increased physical activity or metabolic stress.

High O₂ Levels Increase O₂ Binding

[Lots of O₂-tanks cause up-arrow He-man-globe binding](#)

High O₂ levels increases hemoglobin's binding affinity for oxygen, allowing it to pick up oxygen in areas like the alveolus. The heightened affinity facilitates the efficient loading of oxygen onto hemoglobin, ensuring that oxygen can be transported effectively to peripheral tissues. This mechanism is crucial for maintaining adequate oxygen delivery throughout the body.

High Temperature Reduces O₂ Binding

[Hot-flame makes He-man-globe to drop Down-arrow O₂-tank](#)

High temperature also reduces oxygen-binding affinity in hemoglobin as part of the physiological response to hyperthermia. This phenomenon is particularly evident during exercise, where muscle activity generates heat, leading to a rightward shift in the oxygen-hemoglobin dissociation curve. This shift enhances oxygen unloading in metabolically active tissues. Conversely, lower temperatures increase hemoglobin's oxygen affinity, promoting oxygen loading in the lungs. These temperature-induced changes are part of the body's adaptive mechanisms to meet varying oxygen demands.

High CO₂ Levels Reduce O₂ Binding

[Many CO₂-molecules pushing Down-arrow O₂-tanks Off](#)

High carbon dioxide levels reduce hemoglobin's binding affinity for oxygen, allowing it to drop oxygen off at tissues. Because tissues are undergoing cellular metabolism, they produce CO₂ as a byproduct and use O₂ as an electron acceptor.

Low pH Reduces O₂ Binding

[Acidic-lemon with Down-arrow dropping O₂ tanks](#)

Low pH reduces O₂ binding in a condition called acidosis. If respiration levels are low and CO₂ is not being exhaled fast enough, blood becomes acidic and O₂ binding in hemoglobin is reduced.

Myoglobin Binds Oxygen in Muscle

[Mayo-globe held by Muscle-man](#)

Myoglobin is an oxygen-binding protein found exclusively in muscle cells, including skeletal and cardiac muscles. Unlike hemoglobin, which is composed of four subunits, myoglobin consists of a single polypeptide chain. This structure enables myoglobin to bind oxygen molecules with high affinity, facilitating efficient oxygen storage and transport within muscle tissues.