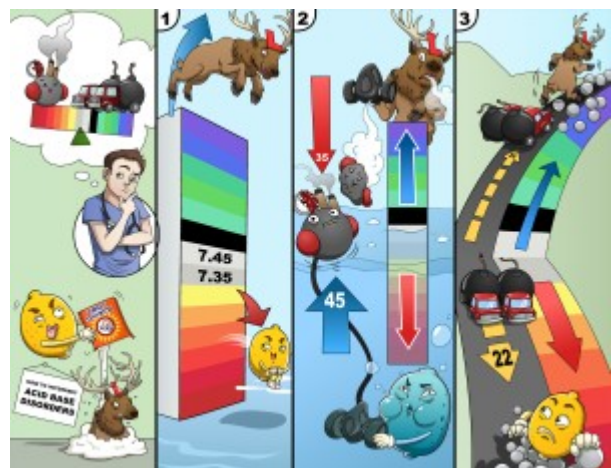


How to Interpret Acid Base Disorders

Interpreting Acid-Base disorders is an essential nursing skill that involves a three-step process: checking the pH, partial pressure of carbon dioxide in the blood (pCO_2), and bicarbonate levels (HCO_3). These indicators will allow you to determine the type of disorder.



PLAY PICMONIC

pH, pCO_2 , HCO_3 (Bicarbonate)

pH-strip, Partial-pressure-gauge CO_2 , and Bi-car-bomb

Mastering quick interpretation of acid-base lab values is a key element to the success of the Nurse. Three components are included in typical lab value assessment: pH, pCO_2 , and HCO_3 (Bicarbonate).

Step 1

pH

pH-strip

The first step to acid-base lab value interpretation is to look at pH. Blood pH is normally 7.35-7.45. pH is determined by the amount of hydrogen ions contained in the blood.

Acidosis

Acidic-lemon

A pH of less than 7.35 is termed acidosis. Acidosis indicates a buildup of carbonic acid in the blood.

Alkalosis

Elk-loser

A pH higher than 7.45 is termed alkalosis. Alkalosis indicates a buildup of bicarbonate (bases) and/or a general decrease in carbonic acid in the blood.

Step 2

pCO_2

Partial-pressure-gauge CO_2

The second step is to examine the partial pressure of carbon dioxide in the blood. pCO_2 is normally 35-45 mmHg and is regulated primarily through respiration.

Opposite Direction as pH

Showing the Opposite Direction on pH-strip

Opposite Respiratory and pH directions indicate a respiratory disorder. If the pCO_2 is not in the opposite direction of the pH, then check the HCO_3 next.

Respiratory Acidosis

[Respirator Acidic-lemon](#)

Respiratory acidosis is often indicated by a pH of less than 7.35 and a $p\text{CO}_2$ of higher than 45 mmHg.

Respiratory Alkalosis

[Respirator Elk-loser](#)

Respiratory Alkalosis is indicated by a pH of more than 7.45 and a $p\text{CO}_2$ of less than 35 mmHg.

Step 3

HCO_3^- (Bicarbonate)

[Bi-car-bomb](#)

The normal value of bicarbonate is 22-26 mmol/L. The amount of the base HCO_3^- , bicarbonate, in the blood is regulated in the kidneys.

Same Direction as pH

[Showing the Same Direction as pH-strip](#)

If the HCO_3^- (bicarbonate) is going in the same direction as pH, then the problem is most likely a metabolic problem.

Metabolic Acidosis

[Metal-ball Acidic-lemon](#)

The patient with Metabolic acidosis can grossly be determined as Down, Down, Down (Decreased pH, Decreased $p\text{CO}_2$, Decreased HCO_3^-).

Metabolic Alkalosis

[Metal-ball Elk-loser](#)

Metabolic alkalosis can grossly be determined as UP, UP, UP (Increased pH, Increased $p\text{CO}_2$, Increased HCO_3^-).